



# Antioxidant and inhibitory activities of $\alpha$ -amylase, $\alpha$ -glucosidase, and G6Pase from *Smallanthus sonchifolius* tuber extract as a potential antidiabetic agent

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## ABSTRACT

**Introduction:** Diabetes mellitus (DM) is a chronic condition with free radicals and carbohydrate-hydrolyzing enzymes playing key roles in its progression. Yacon or *Smallanthus sonchifolius* (Poepp.) H. Rob, a low-sugar crop, has shown promising bioactivities. This research aimed to explore the antidiabetic and antioxidant potential of yacon tuber extract (YTE).

**Methods:** YTE's antioxidant and enzyme inhibitory activities were assessed using 2,2-diphenyl-1-picrylhydrazyl (DPPH) at concentrations of 6.25, 12.5, 25, 50, 100, 200  $\mu\text{g}/\text{mL}$ , hydrogen peroxide ( $\text{H}_2\text{O}_2$ ) scavenging activity at 12.5, 25, 50, 100, 200, 400  $\mu\text{g}/\text{mL}$ , 3-ethylbenzothiazoline-6-sulfonic acid (ABTS) and ferric reducing antioxidant power (FRAP) at 1.56, 3.13, 6.25, 12.5, 25, 50  $\mu\text{g}/\text{mL}$ . The inhibition of  $\alpha$ -amylase,  $\alpha$ -glucosidase (6.25, 12.5, 25, 50, 100, 200  $\mu\text{g}/\text{mL}$ ), and glucose 6-phosphatase (G6Pase) (5.51, 11.03, 22.6, 44.12, 88.24, 176.47  $\mu\text{g}/\text{mL}$ ) was also assessed. Phytochemical analysis identified key bioactive compounds, and  $\text{IC}_{50}$  values were determined to quantify YTE's antioxidant and antidiabetic potentials.

**Results:** YTE contained flavonoids, terpenoids, triterpenoids, and phenolic compounds. It showed highest antioxidant activity ( $P < 0.05$ ) in DPPH, ABTS,  $\text{H}_2\text{O}_2$ , and FRAP assays at 200, 50, 400, and 50  $\mu\text{g}/\text{mL}$ , respectively, compared to the other concentrations. In DPPH,  $\text{H}_2\text{O}_2$ , and ABTS assays, the  $\text{IC}_{50}$  values were 105.77  $\mu\text{g}/\text{mL}$ , 726.64  $\mu\text{g}/\text{mL}$ , and 61.03  $\mu\text{g}/\text{mL}$ , respectively. FRAP activity was 338.68  $\mu\text{M}/\mu\text{g Fe (II)}$  at 50  $\mu\text{g}/\text{mL}$ . YTE also inhibited  $\alpha$ -amylase,  $\alpha$ -glucosidase, and G6Pase with  $\text{IC}_{50}$  values of 174.95  $\mu\text{g}/\text{mL}$ , 222.17  $\mu\text{g}/\text{mL}$ , and 112.51  $\mu\text{g}/\text{mL}$ , respectively.

**Conclusion:** YTE exhibited significant antioxidant properties and inhibited carbohydrate-hydrolyzing enzymes, indicating its potential as an antidiabetic agent.

### Implication for health policy/practice/research/medical education:

This study reported the potential for the anti-diabetic activity of yacon (*Smallanthus sonchifolius*) tuber extract through inhibiting corresponding enzymes. These results might serve as the foundation for further research and development of anti-diabetic medications.

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## Introduction

Diabetes mellitus (DM) is a metabolic condition determined by hyperglycemic symptoms arising from secretory disorders (1,2). Diabetes is usually followed by a rise in oxidative stress, which can affect complications

(3). In patients with hyperglycemia, increased free radical levels are responsible for oxidative stress (4). Reactive oxygen species (ROS) may contribute to type 2 diabetes mellitus (T2DM) by attacking healthy cells and destroying their functional integrity (5).

Carbohydrate-hydrolyzing enzymes, like  $\alpha$ -glucosidase, glucose-6-phosphatase (G6Pase), and  $\alpha$ -amylase aid in starch degradation, as well as the glycogenesis and gluconeogenesis pathways. In starch degradation,  $\alpha$ -amylase and  $\alpha$ -glucosidase enzymes catalyse the hydrolysis of 1,4- $\alpha$ -glycosidic carbohydrate bonds and release  $\alpha$ -glucose. Moreover, a prior study reported that elevated activity of G6Pase leads to an accumulation of endogenous glucose production in type-2 diabetes subjects. The escalated activity of G6Pase initiates a rise in blood glucose concentration, enhancing long-term glycemic control (6). The activities of those enzymes can increase and alter blood glucose levels (5,7,8). Therefore, free radicals' scavengers and inhibition of carbohydrate-hydrolyzing enzymes are needed in DM treatment. In recent years, the capacity of enzymes to defend against free radicals has been improved by antioxidant compounds. These compounds may also minimize oxidative damage and prevent complications of diabetes from progressing. Many anti-diabetic drugs, like biguanides and sulfonylureas, are used for DM treatment. Despite that, these therapies have side effects (9,10).

Medicinal plants are expected to minimize the use of synthetic drugs, which have many side effects. Some medical plants like rambutan, rosella flowers, and white saffron have been used as alternative treatments for various diseases as they are suggested to be rich sources for controlling blood glucose. However, it is necessary to develop further research on the properties, mechanisms of action, and safety of using herbal extracts. In current years, considerable attention has been given to the quest for new compounds with hypoglycaemic and antioxidant properties from plant-based materials, which are supposed to act as antidiabetic agents (4,7,11).

Yacon or *Smallanthus sonchifolius* (Poepp.) H. Rob is a possible usable food origin. Yacon is included in the Asteraceae family and is a native plant to the Andes mountains. One of the benefits of the yacon plant is its antioxidant properties. Researchers are interested in yacon because of its properties and low sugar content (12). Several experiments on yacon have been conducted in search of new functional food, rich in chlorogenic, ferulic, and caffeic acids (13). Some previous research findings indicate that a notable number of phenolic components are found in yacon leaves and yacon tubers (14). Yacon tubers are also found to contain fructo-oligosaccharides (FOS), which are phenolic compounds low in sugar (bioactive compounds) (13). It has antioxidant and antidiabetic properties (15,16); however, its mechanism actions are not clear. Therefore, we evaluated the potential of yacon tuber extract's (YTE's) antidiabetic and antioxidant properties through inhibiting carbohydrate-hydrolyzing enzymes, namely  $\alpha$ -glucosidase,  $\alpha$ -amylase, and G6Pase, as well as antioxidant features by measuring free radical scavenging activities through 2,2-diphenyl-1-picrylhydrazyl (DPPH),

hydrogen peroxide ( $H_2O_2$ ), ferric reducing antioxidant power (FRAP), and 3-ethylbenzothiazoline-6-sulfonic acid (ABTS) assays.

## Materials and Methods

Yacon tuber was gathered from the grow site in Lembang, West Java, Indonesia, and authenticated by Mr. Djuandi a herbarium personnel from the Biology Department, School of Life Science and Technology, Bandung Institute of Technology as *Polymnia edulis* Wedd or *S sonchifolius* (Poepp.) H. Rob. A specimen was deposited there (0220419-A008). The extraction was processed using the maceration method. The tubers were dried, mashed, and extracted using 1500 mL ethanol 70% (Merck 480605). Filtering was repeated every day until the colorless filtrate was obtained. The filtrate was then evaporated to attain YTE. YTE was kept at  $-20\text{ }^\circ\text{C}$  (8,16,17).

The altered Farnsworth procedure was used to classify steroids or triterpenoids, phenols, tannins, saponins, alkaloids, terpenoids, and other flavonoids as the basis for the phytochemical screening of YTE (5,8).

### Phenolics identification

Approximately 0.01 g of sample filtrate was saved on the spotting plate, and 1% of  $FeCl_3$  (Merck 103943) was subsequently added. The coloration in black/red/green/blue/purple showed the presence of phenolic materials (5,7).

### Saponin identification

Approximately 0.01 g of the sample was inserted into a test tube with some water, subsequently simmered for around 5 minutes, and homogenized. If froth was on the surface, the solution was positive for saponins (5,8).

### Steroid/triterpenoids identification

A sample of 0.01 g on the spotting plate was added with a drop of acetic acid and then soaked and closed for 10-15 minutes. A drop of sulfuric acid ( $H_2SO_4$ ) (Merck 1007310510) was included in the sample. Sedimentation with a color of green or blue showed the presence of steroids, meanwhile, orange or red sedimentation marked the presence of triterpenoids (5,8).

### Terpenoid identification

Approximately 0.01 g of sample, vanillin (Merck 108510), and  $H_2SO_4$  (Merck 117048) were put into the spotting plate. The presence of terpenoids was shown by the formation of a purple mixture (5,8).

### Tanin identification

A test tube containing 2 mL 2 N HCl (Merck 1003171000) and a 0.01 g sample was heated in a water bath for 30 minutes. After the mixture went through cooling and filtration steps, it was collected and added to n-amyl

alcohol (Merck 1009751000). The purple color specified a positive reaction with tannin (5,8).

#### Flavonoid identification

A sample of approximately 0.01 g Mg (Merck 1058151000) and HCl 2 N (Merck 1003171000) was inserted into the test tube. The solution was incubated (5 to 10 minutes), and then subjected to cooling and filtration. The filtrate was incorporated into amyl alcohol (Merck 1009751000). Red or orange color formation indicated the presence of flavonoid material (5,8).

#### Alkaloid identification

Approximately 0.01 g of the sample was added with 10% ammonia (Merck 109478) and then vaporized with chloroform (Merck 102431) in a water bath. The base layer was collected. Then, 3 drops of 1N HCl (Merck 109970) were added to the first tube after the formation of 2 layers of the solution. The second tube's residue was dissolved in 1N HCl and the residue was added 2 drops of Dragendorff (Merck 102035). The presence of alkaloids was confirmed if the color changed to yellow (5,8).

#### DPPH scavenging activity assay

Samples of 50  $\mu$ L with different concentrations were injected into the wells on the 96-well microplate with 200  $\mu$ L of DPPH solution (Sigma Aldrich D9132) (0.077 mmol/L in methanol). The samples were put into the incubation under dark conditions for 30 minutes at room temperature. The microplate reader was employed to read the absorbance at a wavelength of 517 nm. Methanol (250  $\mu$ L) was used as a blank, and DPPH (250  $\mu$ L) was employed for the negative control (8,17). The measurement of radical scavenging activity (%) was done by the following formula:

$$\% \text{Scavenging} = (A_c - A_s) / A_c \times 100$$

$A_c$  = absorbance of the negative control (sample was not included).

$A_s$  = absorbance of the sample

#### FRAP scavenging activity assay

Ten milliliters of supplemental acetic acid was combined with 10 mL of compatible acetate buffer (pH=3.6, 300 mM) to create the FRAP reagent (12). One milliliter of ferrous chloride hexahydrate (Merck 1.03943.0250) and 1 mL of 2,4,6-Tris (2-pyridyl) -s-triazine (TPTZ) (Sigma Aldrich 368235-7) were added to 1 mL of distilled water. With 142.5 mL of FRAP reagent, different sample rates of 7.5  $\mu$ L (1000, 500, 250, 125, 62.5, 31.25  $\mu$ g/mL) in the well were applied to 96 well plates. Incubation was done for 30 minutes at 37 °C. The value of the absorbance (Multiskan™ GO Spectrophotometer, Thermo Scientific) was calculated at 593 nm by the microplate reader. FeSO<sub>4</sub> was used for normal curve formation from the concentrations of 1.56-

50  $\mu$ g and FeSO<sub>4</sub> was used from the concentration of 1.56-50  $\mu$ g/mL for the standard curve. The sample yield was expressed in Fe (II)  $\mu$ M/ $\mu$ g extract (8).

#### H<sub>2</sub>O<sub>2</sub> scavenging activity assay

H<sub>2</sub>O<sub>2</sub> scavenging activities were evaluated with this method. For each well on the 96-well plate, 3  $\mu$ L 5mM H<sub>2</sub>O<sub>2</sub> (Merck, 1.08597.1000), 60  $\mu$ L sample, and 12  $\mu$ L ferrous ammonium sulfate (1mM) (Merck, 1.03792.1000) were applied. Then, the mixture was incubated for 5 minutes at room temperature. The sample was incubated at room temperature for 10 minutes, after each well was filled with 75  $\mu$ L of 1,10-phenanthroline (Merck, 1.07223.0010) (briefly). The microplate reader was utilized for the scavenging activity's absorbance measurement at a wavelength of 510 nm (5,8). The H<sub>2</sub>O<sub>2</sub> scavenging activity (%) was quantified using the following formula:

$$\% \text{Scavenging} = (A_c - A_s) / A_c \times 100$$

Where  $A_c$  is the control absorbance and  $A_s$  is the sample absorbance.

#### ABTS scavenging activity assay

2,2'-Azino-bis(3-ethylbenzothiazoline-6-sulfonic acid) (ABTS+) (Sigma A1888-2) diammonium salt-free radical test was used to calculate ABTS activities. ABTS+ solution was formed when 4.9 mM potassium persulfate (volume ratio 1:1) and 14 mM ABTS were reacted with 5.5 mM of PBS (pH 7.4) for 12 to 16 hours at room temperature in a dark environment. The diluted solution was then measured at a wavelength of 745 nm until the absorption of the solution was 0.70  $\pm$  0.02. Different 2  $\mu$ L sample levels (50, 25, 12.5, 6.25, 3.125, 1.5625  $\mu$ g/mL) were added to the 96-well microplate. After that, the samples were added to 198  $\mu$ L of ABTS+ and the plates were put into incubation for 6 minutes at 37 °C. The absorbance was calculated at 745 nm (8,15). To measure the ABTS scavenging operation, the following equation was used:

$$\% \text{ABTS Reducing Activity} = (1 - A_s/A_c) \times 100$$

Where  $A_c$  is the control absorbance and  $A_s$  is the sample absorbance.

#### $\alpha$ -Amylase inhibitory activity assay

A modified method was employed for the  $\alpha$ -amylase inhibitory activity assay test (8,11,17). Thirty microliters of dimethyl sulfoxide (DMSO), 30  $\mu$ L sample, and 30  $\mu$ L blank were added to the wells on a 96-well plate. Then, each well (except empty wells) was added with the  $\alpha$ -amylase enzyme (A7595, Sigma Aldrich). At 37 °C, the mixture was put into incubation for 10 minutes. Forty microliters of starch solution was added to each well, and 40  $\mu$ L of sodium phosphate buffer was added to the control well. The prepared samples were put into incubation at 37 °C for 15 minutes. Then, 100  $\mu$ L acid iodine solution was

administered to each well and the measurement was done using a spectrophotometer at a wavelength of 565 nm. The  $\alpha$ -amylase inhibitory activity (%) was calculated using the formula below, with C as absorbance without starch and S as sample absorbance.

$$\% \alpha\text{-Amylase inhibitory Activity} = (C-S) \times 100/C$$

#### $\alpha$ -Glucosidase inhibitory activity assay

Briefly 1.0 mg  $\alpha$ -glucosidase from *Saccharomyces cerevisiae* (Sigma Aldrich, G5003) was mixed with 100 mL phosphate buffer (pH 7.0) containing 200 mg bovine serum albumin (BSA) (Sigma, A7906) for enzymatic preparation. The mixture was diluted using 1/50 phosphate buffer. Then, 25  $\mu$ L of 20 mM p- $\alpha$ -D-glucopyranoside p-nitrophenyl, 45  $\mu$ L of phosphate buffer, and 5  $\mu$ L sample were mixed to produce 5  $\mu$ L DMSO and 25  $\mu$ L enzyme mixture (negative and positive controls). At 37 °C, the microplate was incubated for 30 minutes and then added with 100  $\mu$ L of 2 M Na<sub>2</sub>CO<sub>3</sub>. Measurement was done using the microplate reader (Multiskan Go Reader) at 400 nm (5,8,11). The activity of  $\alpha$ -glucosidase inhibition was calculated using this equation with C as absorbance without sample and S as sample absorbance.

$$\% \alpha\text{-Glucosidase Inhibitory Activity} = (C-S) \times 100/C$$

#### Glucose-6-phosphatase inhibitory activity assay

Glucose-6-phosphatase (G6Pase) inhibitory activity assay was measured by a modified Barginsky method. Around 10  $\mu$ L glucose-6-phosphatase 0.09 U/mg enzyme (Sigma Aldrich K7410) was added to the control, sample, and blank wells. Then, 40  $\mu$ L of buffer (Sigma Aldrich S7899) was administered to the control well, 10  $\mu$ L into the sample well, and 30  $\mu$ L into the blank well. An amount of 30  $\mu$ L sample was added to the blank and sample wells. The well plate was set into incubation for 20 minutes at 37 °C. An amount of 20  $\mu$ L substrate G6Pase 0.1 M was added to the control and sample wells, then incubated again for 15 minutes at 37 °C. Ammonium molybdate 10% (0.4 g ammonium molybdate in 2.45 M H<sub>2</sub>SO<sub>4</sub>) was added. The enzymatic reaction was stopped by a reducing agent (1% menthol in 3% sodium bisulfite) in each well. The absorbance was quantified with a spectrophotometer at 660 nm wavelength (8). The G6Pase inhibitory activity (%) was calculated using the following equation, with C as absorbance without sample and S as sample absorbance.

$$\% \text{G6Pase Inhibitory Activity} = (C-S) \times 100/C$$

#### Statistical analysis

SPSS program (version 20.0) was used to conduct statistical analyses. The results of DPPH scavenger, H<sub>2</sub>O<sub>2</sub> scavenger, ABTS reduction, and FRAP assays, as well as the  $\alpha$ -amylase,  $\alpha$ -glucosidase, and G6Pase inhibition, were analyzed using linear regression analysis. The linear regression equation that was attained in this step was employed for the Median inhibitory concentration 50

(IC<sub>50</sub>) calculation.

## Results

### Phytochemical screening in yacon tuber extract

The phytochemical screening of YTE detected flavonoids, phenols, triterpenoids, alkaloids, and terpenoids (+), while saponins, tannins, and steroids were undetected (-) (Table 1).

### Antioxidant activity of yacon tuber extract

In this research, the antioxidant activities of YTE were quantified using DPPH, FRAP, H<sub>2</sub>O<sub>2</sub>, and ABTS assays. The results revealed that the scavenging activities of YTE were concentration-dependent or proportional to its concentrations (Figure 1). The results for YTE inhibition activities (IC<sub>50</sub>) in the antioxidant activity test are shown in Table 2. The inhibitory activity (IC<sub>50</sub>) of DPPH radical scavenging in YTE was 105.77  $\mu$ g/mL (Figure 1A). Meanwhile, the FRAP test showed that the most effective YTE concentration for FRAP-reducing activity was 50  $\mu$ g/mL with the scavenging activity of 338.68  $\mu$ M Fe (II) (Figure 1B). The most active YTE concentration for H<sub>2</sub>O<sub>2</sub> scavenging activity was 400  $\mu$ g/mL, with scavenging activity of 28.43% (Figure 1C). Meanwhile, the H<sub>2</sub>O<sub>2</sub> IC<sub>50</sub> value was 726.64  $\mu$ g/mL (Table 2). The highest ABTS scavenging activity from 50  $\mu$ g/mL YTE was 42.29% (Figure 1D). The inhibitory activity (IC<sub>50</sub>) of ABTS radical scavenging activity is shown in Table 2 with the IC<sub>50</sub> value of 61.03  $\mu$ g/mL for YTE.

### Antidiabetic activity assay

The antidiabetic activity of YTE was investigated by testing the inhibitory activity of  $\alpha$ -amylase, G6-Pase, and

**Table 1.** The result of qualitative phytochemical screening of *Smallanthus sonchifolius* extract

Number	Phytochemical test	Results
1	Flavonoid	+
2	Saponins	-
3	Terpenoid	+
4	Triterpenoids/Strenoids	+/-
5	Tannins	-
6	Phenols	+

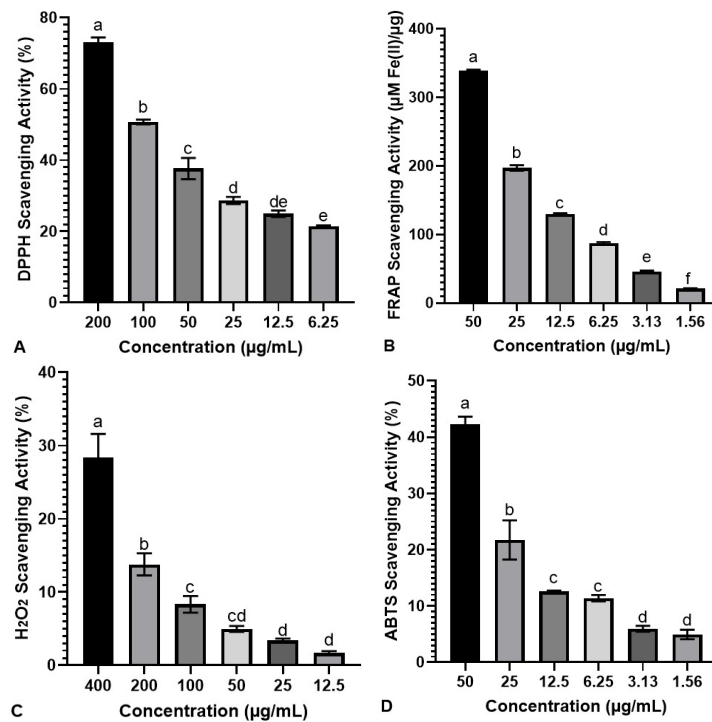
+: Positive results; -: Negative results.

**Table 2.** The antioxidant activity (IC<sub>50</sub>) values of *Smallanthus sonchifolius* extract

Assay	Linear equation	IC <sub>50</sub> Value ( $\mu$ g/mL)
DPPH	Y = 0.2631X + 22.173	105.77
H <sub>2</sub> O <sub>2</sub>	Y = 0.067X + 1.3154	726.64
ABTS	Y = 0.7509X + 4.1718	61.03

DPPH: 2,2-diphenyl-1-picrylhydrazyl; H<sub>2</sub>O<sub>2</sub>: hydrogen peroxide; ABTS: 2,2'-azino-bis-3-ethylbenzothiazoline-6-sulfonic acid.





**Figure 1.** Effects of different *Smallanthus sonchifolius* extract concentrations on antioxidant activities. Data are presented as mean  $\pm$  standard deviation. In each graph, distinct letters demonstrate significant differences among groups as determined by the Tukey HSD post hoc test ( $P < 0.05$ ). A: DPPH (2,2-diphenyl-1-picrylhydrazyl), B: FRAP (ferric reducing antioxidant power), C: H<sub>2</sub>O<sub>2</sub> (hydrogen peroxide), and D: ABTS (2,2'-azino-bis-3-ethylbenzothiazoline-6-sulfonic acid) scavenging activity methods.

$\alpha$ -glucosidase (Figure 2). From the results of this study, it was found that the highest  $\alpha$ -amylase inhibitory activity was achieved with 200  $\mu\text{g/mL}$  YTE, with a result value of 52.7% (Figure 2A). The inhibitory activity ( $\text{IC}_{50}$ ) on the  $\alpha$ -amylase inhibitory activity of YTE was 174.95  $\mu\text{g/mL}$  (Table 3). Meanwhile, the most notable  $\alpha$ -glucosidase inhibitory activity was achieved with 200  $\mu\text{g/mL}$  YTE, with a result of 46.84%. The  $\text{IC}_{50}$  of the  $\alpha$ -glucosidase inhibitory activity of YTE was 222.17  $\mu\text{g/mL}$  (Table 3). Moreover, the most significant G6-Pase inhibition activity was achieved from 176.47  $\mu\text{g/mL}$  YTE (Figure 2C). The  $\text{IC}_{50}$  of  $\alpha$ -glucosidase inhibitory activity is shown in Table 3. In this study, the obtained G6-Pase  $\text{IC}_{50}$  value of YTE was 112.51  $\mu\text{g/mL}$ .

## Discussion

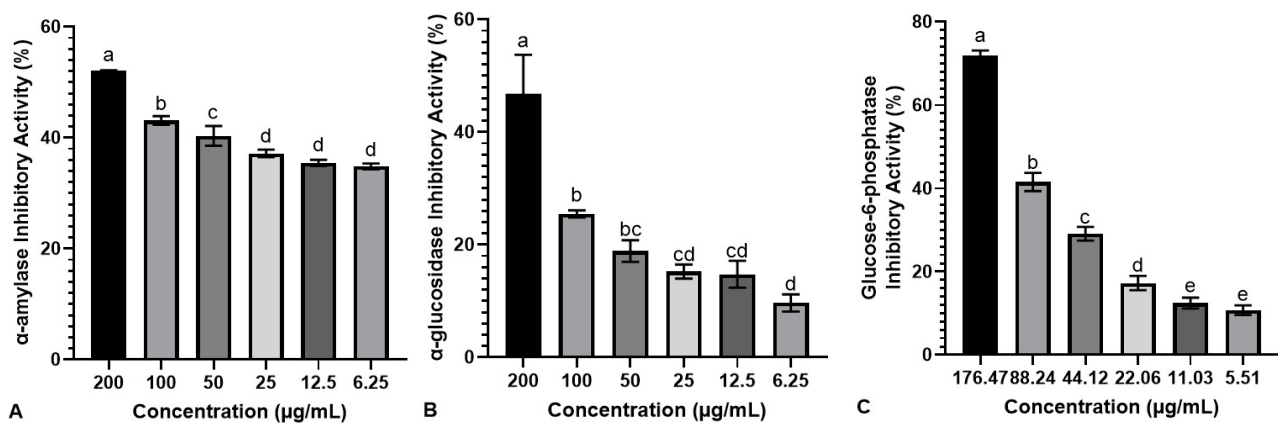
Based on the study, YTE contained flavonoids. This output is consistent with previous studies that evaluated total flavonoid content in meat, skin, and all yacon tubers and cultivars (13). The total phenolic spectrum of

different genotypes (New Zealand, Ecuador, Bolivia, and Germany) was 34.94–68.49  $\text{mg/g}$  (18). Important phenolic acids are found in the tuber, which have the potential for antioxidant, anti-diabetes, and anti-cancer activities. In previous studies, fresh yacon tubers contained phenolic compounds, chlorogenic, and ferulic acid at 46.38, 21.22, and 7.28  $\text{mg/kg}$ , respectively. These studies have also shown that the tubers contained caffeic acid. In comparison to the analysis performed by Pereira et al, ferulic acid and chlorogenic acid concentrations in yacon tubers were found to be very high (2.9  $\text{mg/kg}$  and 39.1  $\text{mg/kg}$ , respectively), with only 27.0  $\text{mg/kg}$  for caffeic acid (19). Differences between yacon tubers' phenolic acid concentrations can be caused by variances between soil type, genotype diversity, climatic conditions, fertilizers, post-harvest time, and agronomic properties (19). Hydroxyl groups in the phenolic compounds contribute to their antioxidant activity and scavenging capacity. Phenolics can also suppress  $\alpha$ -amylase, which will lower the amount of blood glucose (15). Thus, plant extracts with phenolic and flavonoid compounds have many beneficial effects on health such as antioxidant activities. Many bioactive compounds have different hypoglycemic effects as they can come from various plants to inhibit  $\alpha$ -amylase (mostly phenolic, triterpenoids, and flavonoids) and also as antidiabetic agents (15).

Alkaloids can decline blood glucose levels by hindering the action of  $\alpha$ -glucosidase in the duodenal mucosa. The

**Table 3.** The enzyme inhibitory ( $\text{IC}_{50}$ ) value of *Smallanthus sonchifolius* extract

Assay	Linear equation	$\text{IC}_{50}$ value ( $\mu\text{g/mL}$ )
$\alpha$ -glucosidase	$y = 0.0868x + 34.814$	174.95
$\alpha$ -amylase	$y = 0.1822x + 9.5207$	222.17
G6-Pase	$y = 0.3566x + 9.8789$	112.51



**Figure 2.** Effects of different *Smalanthus sonchifolius* extract concentrations on diabetic activities. A: α-amylase, B: α-glucosidase, and C: glucose 6-phosphatase inhibitory activity assays. Data are presented as mean ± standard deviation. In each graph, distinct letters demonstrate significant differences among groups as determined by the Tukey HSD post hoc test ( $P < 0.05$ ).

release of glucose is slowed and its absorption rate into the blood is reduced, preventing peak blood sugar levels (18). Terpenoids can lower blood glucose levels by inducing insulin release and by inducing glucose transporter protein type-4 (GLUT-4) in cells for glucose absorption (20). Meanwhile, flavonoids reduce blood glucose levels by triggering insulin secretion and restoring insulin receptor sensitivity in cells (20). Previous research findings showed that yacon tuber contains derivatives such as coumaric acid, ferulic acid, caffeic acid, and chlorogenic acid (13). This is consistent with previous reports suggesting plant foods containing phenolic compounds might be used as antibacterial, antidiabetic, anti-carcinogenic, and anti-lipidemic agents (21).

In the pathogenesis of DM, several studies have shown the antioxidant and free radical scavenging activities of phenolic compounds. Polyphenols and FOS are found in yacon tubers. These substances affect blood sugar by reducing its levels. As studies have stated, oxidative stress also contributes to blood sugar levels. In yacon tuber, the antioxidant activities reduce oxidative stress, resulting in a decrease in blood sugar levels (1).

DPPH is a stable free radical reagent employed to investigate the antioxidant activity of plant extracts and fractions (22). Antioxidants usually destroy free radicals in DPPH through hydrogen donation (13). The reduction in the conjugated double bonds in DPPH causes the colour to change from violet to yellow (22). The change is measured using a spectrophotometer. According to Figure 1A, the scavenging activity of DPPH radicals was dependent on concentration. As shown by another study, flavonoids can scavenge radicals (23). Antioxidant activity is also calculated by the percentage of inhibition or the amount of antioxidant activity that can capture DPPH free radicals. Higher antioxidant activity is distinct by a lower  $IC_{50}$  value. The  $IC_{50}$  values are categorized into several groups.  $IC_{50} < 50$  µg/mL falls into the “very strong”

antioxidant category,  $IC_{50}$  at values in the array of 50-100 µg/mL fall into the “strong” category,  $IC_{50}$  at values in the range of 101-250 µg/mL fall into the “moderate” category, and  $IC_{50}$  values in the array of 250-500 µg/mL fall into the “weak” category (24). In this study, the  $IC_{50}$  value (105.77 µg/mL) of YTE is categorized as a moderate antioxidant with an  $IC_{50}$  value of 101-250 µg/mL.

Based on a previous study, the DPPH result of yacon’s ethanolic extract was revealed to be 466.92 µg/mL (25). Moreover, studies regarding *S. sonchifolius* with numerous solvents such as n-hexane, chloroform, chloroform/methanol, and methanol reported  $IC_{50}$  values of DPPH scavenging activity within the range of 2.08 and 4.39 µg/mL (15). In the tuber of yacon, chlorogenic acid is the main antioxidant compound. Chlorogenic acid and ferulic acid in yacon tubers are polyphenols, highly related to DPPH radical scavenging (26). Yacon tuber contains polyphenols (2030 mg/kg) with chlorogenic acid predominating at  $48.5 \pm 12.9$  mg/kg (27).

The FRAP procedure is hinged on the decline of the ferriox analog, the  $Fe^{3+}$  complex of tripyridyltriazine  $Fe(TPTZ)^{3+}$  to the strongly blue coloured  $Fe^{2+}$  complex  $Fe(TPTZ)_2^+$  by antioxidants reduction in an acidic medium. In this method, the antioxidant strength is measured based on the absorbance at 593 nm (28). The FRAP test results are equivalent to the iron ion concentration (mM) which can form a complex with the caffeic acid in yacon tubers. This study found that FRAP activity was proportional to the concentration (28). The highest YTE concentration for FRAP-reducing activity was 50 µg/mL, equivalent to 338.68 µM  $Fe(II)$ . A prior study reported that a yacon’s dried extract demonstrated FRAP activity within 31.55 and 66.80 mg TE/g (29).

Another factor that can induce oxidative stress is the presence of  $H_2O_2$ . This compound is a highly reactive oxidizing agent in the presence of metal ions in the human body, leading to sequential oxidations.  $H_2O_2$  can directly

inactivate some enzymes via oxidation of the essential thiol groups (-SH).  $H_2O_2$  can respond to the presence of  $Cu^{2+}$  and  $Fe^{2+}$  ions and form hydroxyl radicals ( $OH\cdot$ ) while passing through the cell membrane. The scavenging of  $H_2O_2$  by phenolic compounds is an important antioxidant defense mechanism. This is caused by their electron-donating ability (30).

Phenolic compounds such as flavonoids can decrease the quantity of ROS and increase the breakdown of  $H_2O_2$  into oxygen ( $O_2$ ) and water ( $H_2O$ ) by the catalase enzyme. Thus, damage to cells and their growth are prevented (31). Based on Table 2, the  $IC_{50}$  of YTE's  $H_2O_2$  scavenging activities in this study is categorized as a frail activity (24).  $H_2O_2$  scavenging activity studies are still limited; however, a previous study showed that yacon extract could significantly increase superoxide dismutase (SOD) antioxidant activity (32). Studies have shown that the extract of yacon with various solvents yielded SOD activity with  $IC_{50}$  values of 0.81-3.81 mg/mL (15). SOD is an enzymatic antioxidant that catalytically converts the superoxide anion ( $O_2\cdot^-$ ) into  $H_2O_2$  and produces  $OH\cdot$  free radicals (30). Studies have shown that yacon extracted with various solvents has SOD activity with  $IC_{50}$  values ranging from 0.81 to 3.81 mg/mL (15).

ABTS is a radical compound containing nitrogen atoms. YTE contains tryptophan, caffeic acid, and chlorogenic acid, which are highly sensitive or reactive to ABTS (26). An ABTS activity test will assess the relative capacity of an antioxidant to catch the resulting ABTS. The blue-green colour is an antioxidant activity marker; hence, ABTS will become colorless (33,34). The blue-green colour reduction is measured at 734 nm (34). This study found that the scavenging activity of ABTS was concentration-dependent. In this study, the  $IC_{50}$  value of YTE (61.03  $\mu$ g/mL) can be classified as a powerful antioxidant (50-100  $\mu$ g/mL) (24). This is supported by previous research that found numerous sections of yacon had ABTS values of 10.38-8,456.2  $\mu$ mol/g TE (19).

The  $\alpha$ -amylase enzyme catalyses the initial step of starch hydrolysis in the digestive system. Inhibiting this enzyme decreases glucose absorbance and can delay starch digestion (7,31). It also participates in lowering post-prandial blood glucose levels. The activity of  $\alpha$ -amylase indicates that the inhibitory activity can increase with rising concentration. This inhibitory activity is affected by phenolic and flavonoid compounds (9). In this study, the  $IC_{50}$  value of YTE  $\alpha$ -amylase inhibitory activity can be categorized as moderate activity (in the range between 101 and 250  $\mu$ g/mL) (24). This finding is supported by a previous study that evaluated the  $\alpha$ -amylase inhibitory activity of yacon that was put into extraction with various solvents (15).

Similar to  $\alpha$ -amylase, the  $\alpha$ -glucosidase enzyme catalyses starch hydrolysis and plays a role in the digestive system. This enzyme cleaves the glycosidic bonds and the

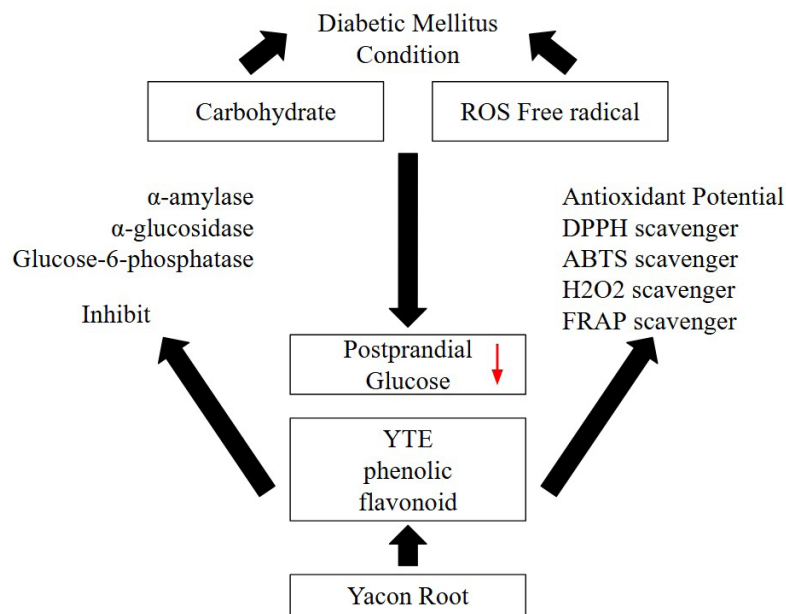
hydrolysis substrate to produce yellow p-nitrophenyl and  $\alpha$ -D-glucose. The  $\alpha$ -glucosidase inhibitor is adequately elevated in samples (35).  $\alpha$ -Glucosidase inhibitor can decrease postprandial plasma glucose levels and modulate post-prandial hyperglycaemia (PPHG) by decelerating the release of D-glucose from complex carbohydrates and detaining glucose absorption (36). This study found that YTE  $\alpha$ -glucosidase inhibitory activity was proportional to concentration and might normalize blood glucose levels. In the current study,  $\alpha$ -glucosidase inhibitory activity had an  $IC_{50}$  value that could be categorized as a moderate antioxidant (24). This inhibition was brought by smaditerpenic acid, which was reported to be similar to acarbose and capable of inhibiting  $\alpha$ -glucosidase. Yacon extract with various solvents was stated to possess  $\alpha$ -glucosidase inhibitory activity with  $IC_{50}$  of 1.00-6.50 mg/mL and concentrations of smallanthaditerpenic acids A (1.43  $\mu$ M), smallanthaditerpenic acids B (1.76  $\mu$ M), and smallanthaditerpenic acids C (1.86  $\mu$ M) (11,37).

The G6Pase enzyme plays a role in regulating glucose homeostasis by hydrolyzing G6P. This hydrolysis produces glucose, which is released into the blood. Inhibiting this enzyme decreases glucose concentration in the blood of DM patients. The G6Pase inhibitory activity assay determines phosphate as the catalytic substrate of the enzyme. Phosphate binds with ammonium molybdate, which in turn binds to the reducing agent (38).

Free phosphates, such as blue phosphomolybdate, determine the level of enzyme activity. YTE glucose-6-phosphate inhibition had an  $IC_{50}$  value of YTE that could be categorized as a moderate activity with an  $IC_{50}$  value of 101-250  $\mu$ g/mL (24). Previous studies have shown that supplementation with fermented yacon leaf tea water extract reduces G6Pase in DM model mice. DM mice had 80.58 nmol/min/mg protein, and the extracts decreased it to 70.39 when supplemented at a low dose and 56.79 nmol/minute/mg when supplemented at a high dose (39).

Yacon tubers contain chlorogenic acid that acts as a competitive inhibitor of glucose 6-phosphate, decreasing its activity (40). The results revealed that the polyphenols in YTE are closely correlated with DPPH, ABTS,  $H_2O_2$ , and FRAP free radical scavenging activities. These correlations highlight their potential antioxidant properties, which play a significant role in anti-diabetes mellitus effects (19,21). Based on these results and the study literature, we proposed a pathway for YTE as an antidiabetic agent (Figure 3).

YTE contains phytochemicals, including phenolics, flavonoids, terpenoids, and alkaloids. Those compounds showed antioxidant activities, including scavenging many oxidants (DPPH, ABTS,  $H_2O_2$ , and FRAP) and inhibiting many glucose-regulating enzymes, such as  $\alpha$ -amylase, G6Pase, and  $\alpha$ -glucosidase. Thus, YTE has the potential to be used as an antidiabetic remedy.



**Figure 3.** Proposed mechanism of *Smallanthus sonchifolius* extract (YTE) as an antidiabetic agent. ROS: Reactive oxygen species.

### Conclusion

YTE was found to have a promising prospect of being used as an antioxidant and antidiabetic agent, probably due to its DPPH, H<sub>2</sub>O<sub>2</sub>, ABTS reducing, and FRAP activities, and carbohydrate-hydrolyzing enzymes (α-amylase, α-glucosidase, and G6Pase) inhibitory activities.

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### Conflict of interests

The authors declared no conflict of interest.

### Ethical considerations

This study does not involve in vivo experiments, and therefore ethical approval is not applicable.

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